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THE ISOMERIZATION/CHLORINATION OF O,O-DIALKYL N,N-DIALKYL THIOPHOSPHORAMIDATE WITH PHOSPHORUS OXYCHLORIDE—A NEW METHOD FOR SYNTHESIS OF S-ALKYL N,N-DIALKYL THIOPHOSPHORAMIDIC ACID DERIVATIVES

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THE ISOMERIZATION/CHLORINATION OF O,O-DIALKYL N,N-DIALKYL THIOPHOSPHORAMIDATE WITH PHOSPHORUS OXYCHLORIDE—A NEW METHOD FOR SYNTHESIS OF S-ALKYL N,N-DIALKYL THIOPHOSPHORAMIDIC ACID DERIVATIVES

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A new convenient method for synthesis of S-alkyl N,N-dialkyl thiophosphoramidic acid derivatives is reported. The chlorination of O,O-dialkyl N,N-dialkyl thiophosphoramidates with phosphorus oxychloride proceeds with isomerization to give S-alkyl N,N-dialkyl thiophosphoramidochloridates, which react further with various nucleophiles in the presence of base to give the title compounds. However, in the reaction of O,O-diethyl N,N-dimethyl thiophosphoramidate with phosphorus oxychloride, an unexpected exchange between chlorine atom and dimethylamino group is observed.

Key words: Thiophosphoramidate, thiophosphorochloridate, isomerization, chlorination.

In the previous paper,^{1,2} we reported that when O-aryl O,O-dialkyl thiophosphates are chlorinated with phosphorus oxychloride, the isomerization of P=S to P—S bond occurs at the same time. This isomerization/chlorination can convert an achiral phosphorus atom into a chiral phosphorus atom to give S-alkyl O-aryl thiophosphorochloridates, which react further with various nucleophiles in the presence of a base to obtain biologically active S-alkyl O-aryl thiophosphoric acid derivatives. Thus, a new convenient method was provided for synthesis of chiral S-alkyl O-aryl thiophosphoric acid derivatives.

TABLE I
 Compounds **2** and **5** prepared

Pro- duct	Yield (%) ^a	bp(°C)/Pa or mp(°C)	n _D ²⁵	Elemental Analysis					
				C%		H%		N%	
				Calc.	Found	Calc.	Found	Calc.	Found
2c	51.8	86–88/40	1.4950	36.60	36.38	7.46	7.29	6.10	5.97
2e	54.8		1.4949	39.43	39.29	7.86	7.57	5.75	5.46
5a	46.3		1.5208	50.95	50.66	7.00	6.95	5.40	5.19
5b	54.5	75–77/27	1.4933	39.80	39.48	8.59	8.67	6.63	6.67
5c	54.9		1.5236	52.73	53.02	7.38	7.30	5.13	4.96
5d	56.1	37–39		36.72	36.38	8.73	8.44	14.27	14.07
5e	67.6	40–42		45.41	45.48	5.99	6.09	3.78	3.70
5f	67.5		1.5424	46.98	46.96	6.37	6.58	8.43	8.15
5g	71.4		1.5082	55.79	55.79	8.03	7.89	4.65	4.45
5h	73.0		1.5043	57.12	57.25	8.31	8.33	4.44	4.49

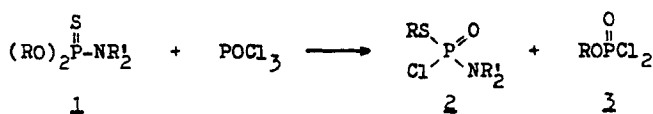
^aTotal yield of two-step reactions based on **1** except **2c** and **2e**.

TABLE II
 IR and ^1H NMR data of compounds 2 and 5

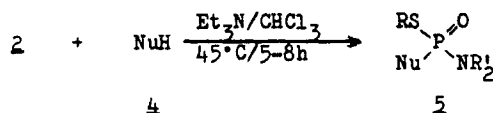
Pro- duct	IR (film or KBr), $\nu(\text{cm}^{-1})$			^1H NMR (CDCl_3/TMS), $\delta(\text{ppm})$, $J_{\text{PH}}(\text{Hz})$
	P-S-C	P=O	P-O-Ar	
<u>2c</u>	575	1240		0.95(t, 3H, CH_3), 1.10(t, 6H, 2CH_3), 1.68(m, 2H, CH_2), 2.75(dt, 2H, $J=16.4$, CH_2S), 3.11(dq, 4H, $J=14.8$, $2\text{CH}_2\text{N}$).
<u>2e</u>	571	1235		0.82(t, 6H, 2CH_3), 1.32(t, 3H, CH_3), 1.50(m, 2H, CH_2), 2.75(dq, 2H, $J=14.4$, CH_2S), 2.97(dt, 4H, $J=12.4$, $2\text{CH}_2\text{N}$).
<u>5a</u>	555	1240	1190 945	1.09(t, 6H, 2CH_3), 2.18(d, 3H, $J=14.8$, CH_2S), 3.18(dq, 4H, $J=12.8$, $2\text{CH}_2\text{N}$), 7.17(s, 5H, Ar).
<u>5b</u>	590	1228		1.14(t, 6H, 2CH_3), 1.35(t, 3H, CH_3), 2.82(dq, 2H, $J=14.4$, CH_2S), 3.17(dq, 4H, $J=12.3$, $2\text{CH}_2\text{N}$), 3.71(d, 3H, $J=12.2$, CH_2S).
<u>5c</u>	595	1235	1190 925	1.15(t, 6H, 2CH_3), 1.36(t, 3H, CH_3), 2.84(dq, 2H, $J=14.8$, CH_2S), 3.26(dq, 4H, $J=12.6$, $2\text{CH}_2\text{N}$), 7.29(s, 5H, Ar).
<u>5d</u>	550	1210		1.09(t, 6H, 2CH_3), 1.29(t, 3H, CH_3), 2.78(m, 2H, CH_2S), 3.12(m, 4H, $2\text{CH}_2\text{N}$), 3.30(s, 2H, NH_2).
<u>5e</u>	566	1231	1197 915	0.78(t, 6H, 2CH_3), 1.15(t, 3H, CH_3), 1.40(m, 2H, CH_2), 2.66(dq, 2H, $J=14.8$, CH_2S), 2.97(dt, 4H, $J=13.0$, $2\text{CH}_2\text{N}$), 6.90-7.50(m, 3H, Ar).
<u>5f</u>	578	1215	1195 949	0.93(t, 3H, CH_3), 1.05(t, 6H, 2CH_3), 1.60(m, 2H, CH_2), 2.67(dt, 2H, $J=14.0$, CH_2S), 3.10(dq, 2H, $J=12.6$, $2\text{CH}_2\text{N}$), 7.60(m, 4H, Ar).
<u>5g</u>	570	1215	1190 945	0.92(t, 3H, CH_3), 1.21(t, 6H, 2CH_3), 1.48(m, 4H, 2CH_2), 2.95(dt, 2H, $J=15.8$, CH_2S), 3.30(dq, 4H, $J=14.4$, $2\text{CH}_2\text{N}$), 7.01(m, 5H, Ar).
<u>5h</u>	560	1242	1190 950	0.91(t, 3H, CH_3), 1.19(t, 6H, 2CH_3), 1.44(m, 4H, 2CH_2), 2.30(s, 3H, CH_3), 2.93(dt, 2H, $J=15.5$, CH_2S), 3.28(dq, 4H, $J=14.4$, $2\text{CH}_2\text{N}$), 7.09(m, 4H, Ar).

Recently, we have found that the isomerization/chlorination of O,O-dialkyl N,N-dialkylthiophosphoramidates 1 with phosphorus oxychlorides can also give the desired products, S-alkyl N,N-dialkyl thiophosphoramidochloridates 2 and O-alkyl phosphorodichloridates 3. The products 2 can be purified by column chromatography on silica gel, e.g. 2c (2; R = Pr, R' = Et), 2e (2; R = Et, R' = Pr) (Tables I and II). Treatment of 2 with nucleophiles NuH(4) in the presence of triethylamine gives S-alkyl N,N-dialkyl thiophosphoramidates 5. The nucleophiles 4 are alcohols, phenols, ammonia and any other compounds with active hydrogen. The crude products 5 are purified by column chromatography on silica gel, distillation under reduced pressure or recrystallization. Thus, eight new compounds 5 have been prepared (Tables I and II).

It is worthy to note that in the reaction of O,O-dialkyl N,N-dimethyl thiophosphoramidate 1f (1; R = Et, R' = Me) with phosphorus oxychloride the desired

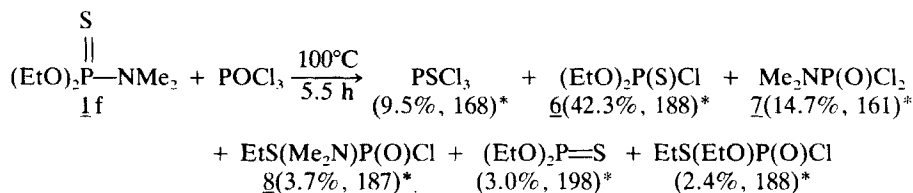


$\underline{1}$	R	R'	t °C	Reaction Time(h)
a	Me	Et	60	4
b	Et	Et	100	6.5
c	Pr	Et	100	10
d	Bu	Et	100	15
e	Et	Pr	100	6



$\underline{5}$	R	R'	Nu	$\underline{5}$	R	R'	Nu
a	Me	Et	PhO	e	Et	Pr	2,4-Cl ₂ C ₆ H ₃ O
b	Et	Et	MeO	f	Pr	Et	4-O ₂ NC ₆ H ₄ O
c	Et	Et	PhO	g	Bu	Et	PhO
d	Et	Et	NH ₂	h	Bu	Et	4-MeC ₆ H ₄ O

product $\underline{2}$ is not obtained, but the exchanged products between chlorine atom and dimethylamino group, namely, O,O-diethyl thiophosphorochloridate $\underline{6}$ and N,N-dimethyl phosphorochloride $\underline{7}$, are formed as the main products. Besides, there are some other by-products. The structures and amounts of seven main components of reaction mixture are determined through GC-MS technique. The amount of each of them is over 2% and their total amount is over 90% (by normalization). In the reaction mixture, material $\underline{1f}$ is undetected, but the amount of phosphorus oxychloride reaches up to 15.3%. The amount of normal isomerization/chlorination product, S-ethyl N,N-dimethyl thiophosphoramidochloridate $\underline{8}$ is only 3.7% of all, but the percentage of the exchanged product $\underline{6}$ is 42%. Compound $\underline{6}$ is obtained by isolation on silica gel column. Thus, the reaction of $\underline{1f}$ with phosphorus oxychloride is shown in the following equation:



In addition, treatment of compounds $(\text{RO})_2\text{P(S)NHR}'$ or $(\text{RO})_2\text{P(S)NH}_2$ (R, R' are same as that of compounds $\underline{1}$) with phosphorus oxychloride under similar con-

*In parentheses are percentages of content and M⁺.

ditions has been investigated in our laboratory. Results show that these reactions are complicated and in some cases very rigorous. The desired isomerization/chlorination products have not been obtained by isolation.

EXPERIMENTAL

1. Apparatus and Reagents

Melting points were determined with a model Yanaco MP-500 apparatus. IR spectra were recorded on a model Shimadzu IR-435 spectrophotometer at film or KBr tablet. ¹HNMR spectra were measured on a JEOL FX-90Q instrument at 90 MHz using TMS as internal standard. GC-MS spectra were recorded on a HP-5890A-GC/5988A-MS instrument. GC conditions are follows: capillary column—Ultra 2, 25 m × 0.32 mm × 0.17 mm film thickness; carrier gas—helium, velocity of flow 1.4 mL/min.; temperature-programmed, injection port temperature 240°C, transfer line temperature 270°C. MS conditions are follows: ion source EI, 200°C, 70 eV; target current 300 μA; scanning range 10–420 amu.

For column chromatography Qingdao silica gel (200–300 mesh) was used. O,O-Dialkyl N,N-dialkyl thiophosphoramidates **1** were prepared by the reaction of O,O-dialkyl thiophosphorochloridates with an excess of dialkylamines using chloroform or acetone as the solvent at 40–45°C for 5–8 h. Phosphorus oxychloride needs to be redistilled before using. Other reagents are commercial.

2. Synthesis of N,N-diethyl S-propyl Thiophosphoramidochloridate **2c**

(Typical procedure): A mixture of 18.6 g (74 mmol) of **1c** (**1**; R = Pr, R' = Et) and 11.3 g (74 mmol) of POCl₃ is heated at 100°C for 10 h with stirring until **1c** disappears (TLC control, solvent system: petroleum ether/Et₂O, 10:1, V/V, iodine as detecting agent). After the removal of the by-product **3** (R = Pr) under vacuum (133 Pa) at 100°C (oil bath), the crude product **2c** is purified by column chromatography on silica gel (200–300 mesh) using petroleum ether/EtOAc (10:1, V/V) as the eluent to give 8.8 g of **2c** with a yield of 52%, bp 86–88°C/39.9 Pa, n_D²⁵ 1.4950 (Tables I and II).

3. Synthesis of N,N-dipropyl S-ethyl O-2,4-dichlorophenyl Thiophosphoramidate **5c**

(Typical procedure): A mixture of 2.5 g (10 mmol) of **1c** (**1**; R = Et, R' = Pr) and 1.5 g (10 mmol) of POCl₃ is heated at 100°C for 6 h with stirring until **1** disappears (TLC control, solvent system: petroleum ether/Et₂O, 10:1, V/V, iodine as detecting agent). After the removal of the by-product **3** (R = Et) under vacuum (266 Pa) at 90°C (oil bath), the residue is dissolved in acetone (10 mL). To the acetone solution is added dropwise a mixture of 10 mL of acetone, 1.6 g (10 mmol) of 2,4-dichlorophenol and 1.5 g (15 mmol) of Et₃N at 20°C, and the mixture is stirred at 45–50°C for 8 h. After cooling to r.t. the reaction mixture is poured into cold water (30 mL). The organic layer is extracted with CHCl₃ (2 × 20 mL). The CHCl₃ layer is washed with water (20 mL), and dried (MgSO₄). After the removal of the solvent the crude product **5c** is purified by column chromatography on silica gel (200–300 mesh) with petroleum ether/EtOAc (10:1, V/V) as the eluent to give 2.5 g of **5c** with a yield of 68%, mp 40–42°C (Tables I and II).

4. The Reaction of O,O-diethyl N,N-dimethyl Thiophosphoramidate **1f** with Phosphorus Oxychloride

A mixture of 7.9 g (40 mmol) of POCl₃ is heated at 100°C for 5.5 h with stirring until **1f** disappears (TLC control, solvent system: petroleum ether/Et₂O, 10:1, V/V, iodine as detecting agent). After the reaction mixture is cooled to r.t., a sample for GC-MS is taken out. To remove a small amount of substance with low boiling point the remainder is distilled at 60°C (oil bath) under reduced pressure (266 Pa) to give 10 g of the crude product, which is purified by using column chromatography on silica gel (200–300 mesh) with petroleum ether/EtOAc (10:1, V/V) as the eluent yielding three fractions. Fraction 1, weighed 4.5 g, n_D²⁵ 1.4670, bp 48–49°C/26.6 Pa, ¹HNMR δ (ppm): 1.38 (t, 6H), 4.15 (dq, 4H, J_{PH} = 10.6 Hz) (Compound **6** in lit.¹; bp 60°C/266 Pa, n_D²⁵ 1.4685), Fraction 2, weighed 1.5 g; Fraction 3, weighed 0.5 g. Fractions 2 and 3 are a complicated mixture.

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